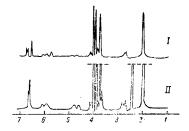
## ALKALOID L-6-A C<sub>(3)</sub> HYDROXYLATED DERIVATIVE OF $\gamma$ -LUMICOLCHICINE

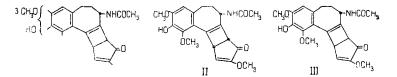
B. Chommadov, M. K. Yusupov, and A. S. Sadykov Khimiya Prirodnykh Soedinenii, Vol. 6, No. 2, p. 275, 1970 UDC 547.944.6

The isolation from <u>Colchicum luteum</u> Baker of two new alkaloids, L-5 and L-6, and the determination of the structure of L-5 as 2-desmethylcolchiceine has been reported previously [1]. The second alkaloid, L-6, with mp 291–293° C  $[\alpha]_D^{20} - 410^\circ$ , has a UV spectrum ( $\lambda_{max}$  228, 264, 282, and 342 m $\mu$ ) resembling those of  $\beta$ - and  $\gamma$ -lumicolchicines. On the basis of its IR and NMR spectra, the developed formula  $C_{15}H_9(OH(OCH_3)_3(CO)(NHCOCH_3))$  was found for it. In addition, the NMR spectrum of the alkaloid L-6 (figure) shows signals ( $\delta$ -scale) of the protons of methylene groups (2.65 and 1.98 ppm), an aromatic proton (singlet, 6.50 ppm), an olefinic proton (doublet at 6.70 ppm), and an imide proton (6.00 ppm), and a methine proton adjacent to nitrogen (4.60 ppm), and also a hydrogen bond (4.08 ppm), which is characteristic for the lumicolchicines [3].



NMR spectrum of alkaloid L-6 (I) and its acetyl derivative (II) (in CDCl<sub>3</sub>).

The mass spectrum of the alkaloid has the peak of the molecular ion  $(M^+)$ , m/e 385 (moderate intensity) and a peak at 342 (very intense). The formation of the second peak is due to the splitting off of an acetyl group from the nitrogen atom, which is characteristic of lumicolchicines [4]. The results of a comparison of the mass spectra of the methyl ether of alkaloid L-6 (mp 275-276° C) and of  $\gamma$ -lumicolchicine shows that they are identical; a mixture gave no depression of the melting point (275-276° C). Consequently, alkaloid L-6 is a previously undescribed derivative of  $\gamma$ -lumicolchicine with a free hydroxyl (I).



In the NMR spectra of alkaloid L-6 and 3-desmethyl- $\beta$ -lumicolchicine (II) the signals of the aromatic protons of ring A are identical, which shows their low sensitivity to changes in the configurations of rings C and D. It is known [5] that a O-acetyl group in the C<sub>(3)</sub> position of ring A, in contrast to positions C<sub>(2)</sub> and C<sub>(4)</sub> shifts the signal of the proton on the C<sub>(1)</sub> atom in the weak-field direction to a relatively small extent. The similar small (~0.10 ppm) shift of this signal into the weak field in acetyl derivatives of II and of the alkaloid L-6 shows that in the latter the hydroxyl group is also present on the C<sub>(3)</sub> atom. On the basis of these facts, for the alkaloid L-6 we propose the structure of 3-desmethyl- $\gamma$ -lumicolchicine (III).

## REFERENCES

- 1. B. Chommadov, M. K. Yusupov, and A. S. Sadykov, KhPS [Chemistry of Natural Compounds], 6, 82, 1970.
- 2. A. Sangster and K. Stuart, Chem. rev., 65, 69, 1965.
- 3. O. L. Chapman, H. G. Smith, and K. W. King, J. Am. Chem. Soc., 85, 803, 1963.
- 4. I. M. Wilson, M. Ohashi, H. Budzikewicz, F. Santavy, and C. Djerassi, Tetrah., 19, 2225, 1963.
- 5. V. Delaroff and P. Rathle, Bull. Soc. chim. France, 6, 1621, 1965.